

**Table F Handout**

name \_\_\_\_\_

List only the precipitates. If both products are aqueous there is NO EVIDENCE that a chemical reaction has taken place, mark those boxes with an "X" for no reaction. Two boxes are your examples.

REACTANT PAIRS	lead (II) nitrate	calcium chloride	potassium hydroxide	silver nitrate	ammonium chloride
sodium chloride	$\text{PbCl}_2(s)$ Lead II chloride	1	2	3	4
sodium sulfate	5	6	7	8	9
sodium hydroxide	10	11	12	14	15
lithium bromide	16	17	18	19	20
potassium phosphate	21	22	23	24	25
ammonium carbonate	26	27	28	29	X no reaction