Answer each question (as if this were a quiz). It's graded for a classwork even though it feels like a quiz. You should be able to do all of these. Put answers on the answer sheet.

- 1. Show the dissociation of solid aluminum chlorate into water.
- 2. Explain why aluminum chlorate is an electrolyte, or is not an electrolyte.
- 3. Determine the number of grams of ammonium chloride that saturates 100 mL solution at 303 Kelvin.
- 4. Determine the number of grams of potassium chloride that saturates a 425 mL solution at 43°C.
- 5. If you have a saturated 100 mL solution of potassium nitrate at 60°C but then cool it down to 20°C, tell exactly what happens? (this is math and words)
- 6. If you have a 100 mL saturated solution of sodium nitrate at 35°C, and you warm it up to 63°C, is it still saturated? Explain, don't say yes or no.
- 7. Draw the dissociation of 1 formula unit of lithium carbonate into water, with the at least 9 water molecules that are properly oriented to each of the ions.
- 8. Water has a low vapor pressure. What is vapor pressure? Why does water have low vapor pressure?
- 9. Ice can float on liquid water (ex: Titanic, the movie). Why is this possible (don't just say ice has a lower density than liquid water, tell WHY water has a lower density than liquid water.
- 10. What is a homogeneous solution? Why won't oil and water mix into a homogeneous solution? What does like dissolves like mean?
- 11. Is barium chromate an electrolyte? If yes, draw a smiley face next to your one word answer. If no, use at least one complete sentence to tell why it isn't.
- 12. Is sodium acetate an electrolyte? No smiley faces here, explain your answer.
- 13. Skip this one.
- 14. Sodium acetate is the ionic compound that is in the reuseable handwarmers. State the particularly important "one-liner" that explains how these hand warmers work.
- 15. Soap is a surfactant. Explain how soap breaks the strong hydrogen bonding at the surface of water.
- 16. 325 grams of steam at 373 Kelvin condenses onto your kitchen window, then cools down to the room temperature of 23°C. What formula or formulas do you need to use to figure this out? You do NOT hve to do the math for this one.

Water In Class Handout Answers

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