

Naming compounds Practice page here—answers on page 2

barium chromate

zinc dichromate

cobalt (II) permanganate

nickel (II) sulfite

potassium phosphide



ammonia

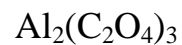
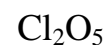
lead (IV) chlorite

ammonium phosphate

silver carbonate

gold (III) fluoride

copper (I) cyanide



diarsenic trisulfide

lead (II) chromate

hydrogen fluoride

sodium hydroxide

cesium hydrogen carbonate

magnesium hydrogen sulfate

barium chromate	BaCrO_4
zinc dichromate	ZnCr_2O_7
cobalt (II) permanganate	$\text{Co}(\text{MnO}_4)_2$
nickel (II) sulfite	NiSO_3
potassium phosphide	K_3P
chromium (III) acetate	$\text{Cr}(\text{C}_2\text{H}_3\text{O}_2)_3$
mercury (II) oxide	HgO
hydrogen sulfide	H_2S
dihydrogen monoxide	H_2O
ammonia	NH_3
lead (IV) chlorite	$\text{Pb}(\text{ClO}_2)_4$
ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
silver carbonate	Ag_2CO_3
gold (III) fluoride	AuF_3
copper (I) cyanide	CuCN
dichlorine pentoxide	Cl_2O_5
chromium (VI) iodide	CrI_6
aluminum oxalate	$\text{Al}_2(\text{C}_2\text{O}_4)_3$
calcium thiocyanate	$\text{Ca}(\text{SCN})_2$
diarsenic trisulfide	As_2S_3
lead (II) chromate	PbCrO_4
hydrogen fluoride	HF
sodium hydroxide	NaOH
cesium hydrogen carbonate	CsHCO_3
magnesium hydrogen sulfate	$\text{Mg}(\text{HSO}_4)_2$