

SR — Atoms + AQ → Different AQ + different atoms Table J

DR — AQ + AQ → Different AQ + solid Table F

For each reaction, change word equation into a balanced chemical equation with phase symbols.  
 Watch out for “no reactions” too — these get balanced too. Write SR or DR or X in top right corner

Potassium metal and zinc chloride solution...	<input type="checkbox"/>
Aluminum metal and silver nitrate solution...	<input type="checkbox"/>
Copper (II) sulfate solution and calcium nitrate solution...	<input type="checkbox"/>
Strontium metal and sodium chloride solution...	<input type="checkbox"/>
Sodium permanganate solution and lithium chromate solution...	<input type="checkbox"/>
Barium metal and potassium hydroxide solution...	<input type="checkbox"/>
Lead (II) nitrate solution and ammonium carbonate solution...	<input type="checkbox"/>
Sodium phosphate solution and cobalt (III) acetate solution...	<input type="checkbox"/>

Ammonium chromate solution and sodium chloride solution..

Aluminum hydrogen carbonate solution and potassium sulfide solution...

Tin (IV) chloride solution and zinc metal...

Magnesium nitrate solution and iron metal...

Iron (III) nitrate solution and magnesium metal...

Silver chlorate solution and sodium chloride solution...

Ammonium sulfate solution and barium acetate solution...

Calcium hydroxide solution and beryllium nitrate solution...

Lithium phosphate and ammonium chromate solutions...