

Naming compounds Practice page here—[answers on page 2](#)

barium chromate	
zinc dichromate	
cobalt (II) permanganate	
nickel (II) sulfite	
potassium phosphide	
	$\text{Cr}(\text{C}_2\text{H}_3\text{O}_2)_3$
	$\text{HgO}$
	$\text{H}_2\text{S}$
	$\text{H}_2\text{O}$
ammonia	
lead (IV) chlorite	
ammonium phosphate	
silver carbonate	
gold (III) fluoride	
copper (I) cyanide	
	$\text{Cl}_2\text{O}_5$
	$\text{CrI}_6$
	$\text{Al}_2(\text{C}_2\text{O}_4)_3$
	$\text{Ca}(\text{SCN})_2$
diarsenic trisulfide	
lead (II) chromate	
hydrogen fluoride	
sodium hydroxide	
cesium hydrogen carbonate	
magnesium hydrogen sulfate	

barium chromate	$\text{BaCrO}_4$
zinc dichromate	$\text{ZnCr}_2\text{O}_7$
cobalt (II) permanganate	$\text{Co}(\text{MnO}_4)_2$
nickel (II) sulfite	$\text{NiSO}_3$
potassium phosphide	$\text{K}_3\text{P}$
chromium (III) acetate	$\text{Cr}(\text{C}_2\text{H}_3\text{O}_2)_3$
mercury (II) oxide	$\text{HgO}$
dihydrogen monosulfide	$\text{H}_2\text{S}$
dihydrogen monoxide	$\text{H}_2\text{O}$
ammonia	$\text{NH}_3$
lead (IV) chlorite	$\text{Pb}(\text{ClO}_2)_4$
ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
silver carbonate	$\text{Ag}_2\text{CO}_3$
gold (III) fluoride	$\text{AuF}_3$
copper (I) cyanide	$\text{CuCN}$
dichlorine pentoxide	$\text{Cl}_2\text{O}_5$
chromium (VI) iodide	$\text{CrI}_6$
aluminum oxalate	$\text{Al}_2(\text{C}_2\text{O}_4)_3$
calcium thiocyanate	$\text{Ca}(\text{SCN})_2$
diarsenic trisulfide	$\text{As}_2\text{S}_3$
lead (II) chromate	$\text{PbCrO}_4$
hydrogen fluoride	$\text{HF}$
sodium hydroxide	$\text{NaOH}$
cesium hydrogen carbonate	$\text{CsHCO}_3$
magnesium hydrogen sulfate	$\text{Mg}(\text{HSO}_4)_2$